

ORISSVICH, N.A.; MALOSKAYA, G.A.

Temperature dependence of the intensity of infrared absorption
bands of the fundamental molecular vibrations in vapors. (pt. 1)
spektr. 15 no.5:772-775 My 1964. (MIRA 1964)

L 3207-66

ACCESSION NR: AR5012249

UR/0058/65/000/003/D031/D031

SOURCE: Ref. zh. Fizika, Abs. 3D224

AUTHORS: Boresevich, N. A.; Zaleskaya, G. A.

TITLE: Investigation of the temperature dependence of the intensities of infrared absorption bands of the fundamental oscillations of molecules in vapors

CITED SOURCE: Tr. Komis. po spektroskopii. AN SSSR, vyp. 1, 1964, 721-730

TOPIC TAGS: temperature dependence, infrared spectrum, absorption band, molecular vibration, vapor

TRANSLATION: A procedure is developed for measuring the infrared absorption spectra of substances that are difficult to sublimate. The temperature dependence of the integral sensitivity of the absorption band of the valence vibrations of groups CH, NH, and NH₂ is investigated. With increasing temperature, a shift of the bands to the long-wave region of the spectrum is observed. The half-width of the bands increases in this case, with simultaneous decrease in the intensity at the maximum. The integral intensity increases with the increasing temperature. See also RZhFiz, 1964, 12D214

SUB CODE: OP, NP
Card 1/1

ENCL: 00

BONISLICH, N.A.; PETROVICH, P.I.; ZALESSKAYA, G.A.

Infrared spectra of n-xylene derivatives. Dokl. AN BSSR 4 no.12:
510-513 D '60. (MIRA 14:2)

1. Institut fiziki AN BSSR i Nauchno-issledovatel'skiy institut
organicheskikh poluproduktov i krasitolei im. Voroshilova.
(Xylene---Spectra)

1. ZALESSKAYA, G.N.
2. USSR (600)
4. RUBTSOVO * LIMESTONE
7. Report on the detailed prospecting for flux limestone in the area of Rubtsovo and Podasnov'ye villages in the Belokholunitskiy District of the Kirov Province in 1944. (Abstract) Izv. Glav, upr. geol. fon. no. 2 1947

9. Monthly List of Russian Accessions, Library of Congress, March, 1953. Unclassified.

1, ULANOV, I. I., ZALFSSKAYA, G. N.

2. USSR (600)

4. Omutninsk District - Fire Clay

7. Report on the detailed survey of the Peskovka and Kokorinskiy refractory clay deposits in the Omutninsk District of the Kirov Province. (Abstract.) *Izv.Glav.upr.geol.* fon. no. 2, 1947

0. Monthly List of Russian Acquisitions. Library of Congress. March 1953. Unclassified.

ZALESSKAYA, L., kand.arkhitektury

Ornamental trees and parks in residential districts. Zhil.stroi.
no.4:24-27 '59. (MIRA 12:6)

(Landscape architecture) (Parks)

USSR/Medicine - Sleep Treatment

Nov 51

"Experiments With Sleep Treatment of Ulcerous Diseases in the District Hospital," A. A. Tchakova, L. A. Zaleskaya, Sormovo Rayon Hosp

"Klin Med" Vol XXIX, No 11, pp 53-55

Expts were made with a number of patients suffering from ulcers. Cortical (chloral hydrate) and subcortical soporifics (medinal, luminal, veronal) were used. To produce deeper sleep, 1-2% bromide was added to the soporific doses. Temp, pulse, respiration, blood pressure were

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USSR/Medicine - Sleep Treatment
(Contd)

Nov 51

checked at regular intervals. Final results were excellent in 10 cases, satisfactory in 11, unsatisfactory in 5. Pain decreased and so did dispeptic symptoms. The general state of health improved. No pathological changes either in the blood or urine were found during or after treatment.

203173

ZALESKAYA, L. A.

TSEL', Ye.A.; ZALESSKAYA, L.I.

Minutes of meetings Nos. 33 and 34 of the Leningrad and Leningrad
Province Oncological Society, November 13 and December 11, 1958.

Vop.onk. 5 no.6:759-763 '59.

(TUMORS)

(MIRA 12:12)

GOL'DSHTEYN, L.M.; BYCHENKOVA, M.N.; ZALESSKAYA, L.I.

Effectiveness of radiotherapy by various methods of cancer of the
upper segments of the esophagus. Trudy Inst.onk.AMN SSSR no.4:27-
38 '62. (MIRA 15:9)

(ESOPHAGUS—CANCER) (RADIOTHERAPY)

DYMARSKIY, L.Yu.; DIL'MAN, V.M.; ZALESSKAYA, L.I.; ZIV, M.A.; BOGIEOV,
Ye.A.; PAVLOVA, M.V.

Combined hormone and chemotherapy and radiotherapy of far
advanced breast cancer. Vop. onk. 9 no.7:44-52 '63.

(MIRA 16:12)

1. Iz Instituta onkologii AMN SSSR (nauchnyy rukovoditel' raboty
chlen-korrespondent AMN SSSR prof. S.A. Kholdin). Adres avtorov:
Leningrad, P-129, Institut onkologii AMN SSSR.

POL'KINA, R.I.; BYCHENKOVA, M.N.; ZALESSKAYA, L.I.

Attempts at the radiosensitization of transplanted tumors using
some pyrimidine derivatives. Trudy Inst.onk.AMN SSSR no.4:102-
107 '62. (MIRA 15:9)

1. Iz laboratorii eksperimental'noy onkologii (zav. - prof. N.V.
Lazarev) i rentgenovskogo otdeleniya (zav. - prof. L.M.Gol'dshteyn).
(PYRIMIDINES) (ONCOLOGY) (RADIOTHERAPY)

ZALESSKAYA, L.I.

Minutes of the 30th session of the Leningrad and Leningrad Province
Society of Oncologists, May 8, 1958. Vop.onk. 4 no.6:755-757 '58.
(MIRA 12:1)

(TUMORS)

Zalesskaya, L. I.

MOSCOW.

Publichnaia biblioteka. Vladimir Vladimirovich Maiakovskii, 1893-1930. Moskva,
1950. 31 p. Its: Sovetskie pisateli 51-20614

Z8542.9.M6

~~ЗАЛЕССКАЯ, Л.И.~~
ZALESSKAYA, L.I.

Minutes of the 20th and 21st sessions of the Society of Oncologists
of Leningrad and Leningrad Province. Vop.onk. 3 no.6:769-773 '57.

(MIRA 11:2)

1. Sekretar' Nauchnogo obshchestva onkologov Leningrada i Lenin-
gradskoy oblasti
(TUMORS)

EXCERPTA MEDICA Sec 16 Vol 7/9 Cancer Sept 59

3869. Combined external and intracavitary radiation treatment of oesophageal cancer (Russian text) ZALESSKAYA L. I. Inst. of Oncol. AMS, Leningrad, USSR *Vopr. Onkol.* 1959, 5/2 (178-183) Tables 1 Illus. 3
The combination of external radiation with roentgen rays through a lead grid and intracavitary radiation with gamma rays Co^{60} was used in 15 patients. Improvement was noted in 13 cases. The effect of treatment proved to be unsteady. After 2-3 months many patients developed pains in the thorax without tumour recurrence and all of them died 5-18 months later.

ZALESKAYA, L.I.

Minutes of the 25th session of the Leningrad and Leningrad Province
Society of Oncologists. Vop.onk. 4 no.3:379-381 '58 (MIRA 11:8)
(TUMORS)

POL'KINA, R.I.; BYCHENKOVA, M.N.; ZALESSKAYA, L.I.

Radiosensitization of inoculated tumors with some pyrimidine derivatives. Vop. onk. 9 no.9:34-38 '63. (MIRA 17:9)

1. Iz laboratorii eksperimental'noy onkologii (zav.- prof. N.V. Lazarev) i rentgenovskogo otdeleniya (zav.- prof. L.M. Gol'dshteyn [deceased]) Instituta onkologii AMN SSSR (dir.- deystvitel'nyy chlen AMN SSSR prof. A.I. Serebrov). Adres avtorov: Leningrad, P-129, 2-ya Berezovaya alleya 3, Institut onkologii AMN SSSR.

S/020/60/132/02/35/067
B011/B002

AUTHORS: Topchiyev, A. V., Academician, Kaptsov, N. N., Zalesskaya, L. N.

TITLE: Nitration of Paraoxydiphenyldimethylmethane Acetate in the Presence of Urea

PERIODICAL: Doklady Akademii nauk SSSR, 1960, Vol. 132, No. 2, pp. 371-373

TEXT: The authors proved that during the nitration of paraoxydiphenyldimethylmethane acetate, one of the three nitro groups enters a non-phenolic cycle of the molecule (see scheme). For the purpose of purification p-oxydiphenyldimethylmethane (ODDM) (commercial by-product of the phenol acetone production) was first recrystallized from a mixture of benzene-petroleum ether. The ODDM crystals are white, needle-shaped and have their melting point at 73°-75°. Production of the acetate: ODDM was dissolved in an aqueous KOH solution with an addition of ethanol, and 180 g of acetic anhydride were quickly added. After it was cooled down for half an hour by adding lumps of ice, or when the mixture was put on ice, the solution separated in layers. It was extracted by means of ether. When the ether was distilled off, the remaining substance was a colorless, thick liquid which could be distilled almost without decomposition at 327° at

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Nitration of Paraoxydiphenyldimethylmethane Acetate
in the Presence of Urea

S/020/60/132/02/35/067
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atmospheric pressure. The melting point of this acetate was 180° - 182° /1.5 mm. The molecular weight was determined to be 250 and calculated to be 254. The acetate easily dissolves in benzene, benzine, o-xylene, and other solvents. Nitration of the acetate by means of HNO_3 at 15° - 20° leads to the formation of picric acid. This can be prevented if the acetate is poured off at lower temperatures and if the reaction mass is left standing at a lower temperature. Thus low yields of a yellowish crystalline substance develop with a melting point of 127° . It was analyzed to be the trinitro derivative of p-ODDM. Its molecular weight was determined to be 356 and calculated to be 347. In order to avoid the oxidative action of HNO_3 , the authors nitrated ODDM acetate in the presence of urea. Table 1 shows that in this case, the trinitro compound develops with a considerably higher yield. Urea however, must be added after the acetate has been poured off, otherwise only picric acid would develop. The position of the nitro groups was proven by oxidation with chromic acid. A small amount (0.07 g) of a solid yellow substance was obtained with a melting point between 238° and 241° . The authors compared it with para-nitro-benzoic acid whose melting point is at 241° (Scheme). There are 1 table and 4 references, 1 of which is Soviet.

Card 2/3

Nitration of Paraoxydiphenyldimethylmethane Acetate
in the Presence of Urea

S/020/60/132/02/35/067
B011/B002

ASSOCIATION: Institut neftekhimicheskogo sinteza Akademii nauk SSSR (Institute
of Petroleum-chemical Synthesis of the Academy of Sciences, USSR) ↙

SUBMITTED: February 17, 1960

Card 3/3

SHKREBEL', M.Ya.. Prinimali uchastiye: BLAGOVESHCHENSKAYA, K.A.;
DZYUBENKO, G.P.; FRAGAYLOVA, V.I.; ZALESSKAYA, L.O.; KOTSERUBA,
L.P.; KOVBASENKO, L.A.; LYAUDANSKAYA, B.Ye.; MILOVZOROV, P.Z.
[deceased]; NEZHUREBDA, M.P.; SNITKO, K.I.; YANTSOVA, A.V..
KRESHCHENSKIY, Ye.S., tekhn.red.

[Economy of Kiev Province; a statistical manual] Narodnoe kho-
zaiatvo Kievskoi oblasti; statisticheskiy sbornik. Kiev, Gos.
stat.isd-vo, 1959. 255 p. (MIRA 13:3)

1. Kiev (Province) Statisticheskoye upravleniye. 2. Nachal'nik
statisticheskogo upravleniya Kiyevskoy oblasti (for Shkrebel').
(Kiev Province--Statistics)

ZALESSKAIA, L.S.

ZALESSKAIA, L.S. ...Ozelenenie gorodov Srednei Azii. Dendrologicheskaiia chast'
F.N. Rusanova. Moskva, 1949. 93, (3) p. (Akademiia arkhitektury SSSR).
"Bibliografiia": p. 92-94.

DLC: SB484.R9Z3

SO: LC, Soviet Geography, Part II, 1951, Unclassified

ZALESSKAYA, L.S., kand.arkh.; ALEKSANDROVA, V.D., arkh.; SHKVARIKOV, V.A., red.; DYURNBAUM, M.S., red. [deceased]; KOLMESNIKOV, A.I., red.; DOMSHLAK, I.P., red.; BALAKSHINA, Ye.S., arkhitektor, red.; FRIDBERG, G.V., inzh., red.; BRUSINA, L.N., tekhn.red.

[Manual for architects] Spravochnik arkhitektora. Red.V.A. Shkvarikov i dr. Moskva, Gos.izd-vo lit-ry po stroit., arkh. i stroit.materialam. Vol.3., pt.2. [Landscaping of cities] Ozenenie gorodov. Sost. L.S.Zalasskaya i V.D.Aleksandrova. 1960. 463 p. (MIRA 13:9)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut gradostroitel'stva i rayonnoy planirovki. (Landscape gardening)

ZALESSESKAYA, L S

Ozeleneniye Stolitzy [Landscaping of the Capital] Moskva, Gos. Izd-vo., Literatury
Po Stroitel'stvu i Arkhitekture, 1953.
39 p. illus. (Arkhitektura Moskvyy, By Tsika Lektsiy)

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ZALESSKAYA, M. A.

Zalesskaya, M. A. and Orlova, M. Ye. "Cytobacterioscopy of cervical canal discharges in gonorrhea", Sbornik nauch. trudov (Rost. obl. nauch.-issled. akushersko-g nekol. in-t), Issue 8, 1948, p. 56-60.

ZALESKAYA, M. A.

Nikol'skiy, V.V, Zaleskaya, M.A. and Chukrayeva, N.I. "The dynamics of RN changes and the albumen content in discharges of gonorrhea patients under the influence of pencillin and sulfadin therapy", Sbornik nauch. trudov (Rost, Obl. nauch.-issled. akushersko-ginekol. in-t.), Issue 8, 1948, p. 72-76.

So: U-3261, 10 April 1953 (Letopis 'Zhurnal 'nykh Statey, No. 12, '949).

ZALESSKAYA, M. A.

Zalesskaya, M. A. "The use of AtsS in combination with sulfamide drugs in the treatment of postnatal septic diseases", Sbornik nauch. trudov (Rost. obl. nauch.-issled. akushersko-ginekolog. in-t), Issue 8, 1948, p. 162-167.

So: U-3261, 10 April 1953 (Letopis 'Zhurnal 'nykh Statey, No. 12, 1949).

Zalesskaya, N.D.

USSR/Forestry - Forest Culture.

J-4

Abs Jour : Referat Zhur - Biologiya, No 16, 25 Aug 1957, 69128

Author : Zalesskaya, N.D.

Inst :

Title : Irrigation of Oaks Planted in Nidi in Field Protective Forest Strips.

Orig Pub : Nauch. tr. Ukr. n.-i. in-ta gidrotekhn. i melior., 1956, No 77/3, 27-32

Abstract : In 1950, in Brilevsk scientific-experimental irrigation station, forest strips were introduced for the study of irrigation effect on the growth and development of oaks. It is established that with necessary load of irrigation and correct and timely soil cultivation, the normal development of oaks is secured and the yearly gain averages 39 to 60 cm. A greater density of oaks in a nidus (up to 20 trees) did not inhibit their development and aided greater height of matures oaks. By the end of the second

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USSR/Forestry - Forest Culture.

J-4

Abs Jour : Referat Zhur - Biologiya, No 16, 25 Aug 1957, 69128

year (while irrigated) a total coverage occurred in nidi.
In the initial years of their life the irrigated small
oaks do not yield in growth to fast-growing species.

Card 2/2

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16

Obtaining higher fusel-oil yields in alcoholic fermenta-
 tion. M. I. Zakrevskaya. *Microbiology* (U. S. S. R.) 6,
 604-615 (1967). *Cibac. Zbir.* 1937, 11, 2837. — An addn.
 of 0.3% leucine is necessary to increase the fusel-oil yield
 in the fermentation of sugar acids. With the optimum
 sugar concn. of 10% a fusel-oil yield of 13.5% of the alc.
 or 0.16% of the mash was obtained. The amino N of the
 leucine was 70-80% assimilated. A 10% sugar soln. with
 ground aeration and the add'n. of 0.5% leucine gave a 0.07%
 fusel-oil yield (0.37% of the mash); a 6% sugar soln. gave
 a yield of 7.5%. Substitution of other amino acids for
 leucine reduced the fusel-oil concn. to 0.08 or even 0.04%.
 M. G. Mayer

ASB-55A METALLURGICAL LITERATURE CLASSIFICATION

6-27-71

1ST AND 2ND COPIES										3RD AND 4TH COPIES									
PROCESS AND PROPERTIES INDEX																			
CA					16														
<p>Formation of fusel oil during alcoholic fermentation. H. M. L. Zelenkaya. <i>Microbiology</i> (U. S. S. R.) 7, No. 6, 646-64 (1938); <i>Khim. Referat. Zhur.</i> 2, No. 4, 66-7 (1939); cf. G. A. 29, 8980. —The use of tech. leucine lowered the yield of fusel oil to 0.12%; with pure leucine the yield was 0.14%. The most energetic formation of fusel oil coincides with the beginning of the main fermentation of sugar, when the multiplication of yeast and the domination of leucine are nearly concluded, in the period of from 24 to 48 hrs. The greater the amt. of the fermented sugar the greater the yield of fusel oil. On lowering the amt. of added yeast (3 million cells/cc.) the formation of fusel oil takes place slowly and continues uniformly until the fermentation is concluded, but the yield is greater (0.23%). With a larger amt. of yeast (30 million cells/cc.) the formation of fusel oil stops rapidly (in 36 hrs.) simultaneously with the main fermentation of sugar and the yield of fusel oil is smaller (0.10%). The splitting of leucine by yeast takes place very rapidly; within 12 to 24 hrs. up to 70% of the leucine is consumed.</p> <p>W. R. Heng</p>																			
ASM-A METALLURGICAL LITERATURE CLASSIFICATION																			
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FROM 510.0114										FROM 500.0114									

1ST AND 2ND GROUPS										3RD AND 4TH GROUPS									
PROCEDURES AND PROPERTIES INDEX																			
CP										16									
<p>Formation of fuel oil in alcoholic fermentation. M. I. Zolotarevskiy. <i>Microbiology</i> (U. S. S. R.) 10, 97-116(1941); cf. C. A. 34, 1123^a.—A review with 60 references. T. Laanes</p>																			
ASB-5LA METALLURGICAL LITERATURE CLASSIFICATION																			
FROM STEELING										FROM ROMANIN									
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Riboflavin formation in *Aspergillus flavus* mycelium grown in filtered cereal mashes. M. I. Zaleskaya (All-Union Research Inst. Ak. Inst. Mikrobiol. SSSR, No. 10, 127-30 (1959)). The solids in cereal-potato mash filtrates are 85% assimilable for *Aspergillus flavus*. Mycelium prepared from the cultures contain 4.5-5.5% N (20-34% protein). Riboflavin content is 30-92 γ g. (calcd. on dry wt.), depending on fermentation time and the solids content of the filtrate. In old cultures riboflavin passes from the mycelium into the medium. J. P. S.

ZALESSKAYA, M. I.

2
Starch hydrolysis in acetone-butanol fermentation. I.
Amylolytic enzymes of acetone-butanol bacteria. M. I.
Zalesskaya and F. M. Klinburgskaya. *Trudy Vsesoyuz.*
Nauka-Issledovatel. Inst. Spirtovoi Prom. 1954, No. 3, 140-
148. Three components were found in the acetone-butanol
bacterial amylase system: amylolytic, strongly dextrinoly-
tic, and saccharifying. Evidence of amylolytic activity in
the culture medium appeared as early as 8 hrs. after the
onset of fermentation and reached its max. after 36-48 hrs.
The addn. of β -amylase enhanced the amylolytic activity in
the culture medium. The presence of maltose in the culture
medium was also demonstrated. II. Changes in the carbo-
hydrate composition of the medium at the different stages
of acetone-butanol fermentation. *Ibid.* 149-155. A 6-8%
rye flour mash was fermented. Saccharification of the starch
was almost complete 12 hrs. after the onset of fermentation.
The sugars thus formed are completely fermented in 30 hrs.
Pentosans remained totally unaffected by the fermentation
processes. Through *Referat. Zhur. Khim., Biol. Khim.*
1955, No. 10712. B. S. Levine

ZALESSKAYA, M.I.

How composition of the medium influences by sol-ace-
tone; frequently M. I. Zaleskaya (All-Union Sci.
Research Inst. for Ind., Moscow). *Mikrobiologiya* 24,
455-51 (1955).—Yield of solvents, and especially of BuOH,
is higher in rye mashies than in corn or wheat mashies, which
are richer in protein than rye. Yield was also raised in corn
mashes (e.g. from 3.48 g./l. to 4.14 for Me₂CO and from
7.76 to 9.626 for BuOH) by increasing the starch:protein
ratio; starch was added to the corn flour to raise starch con-
tent from 60.3 to 68.06%. Yields are also tabulated for
pea:potato, pea:corn, corn:potato, wheat and barley
mashes. Julian P. Smith

ZALESSKAYA, M. I.

7 5

~~Influence of proteolytic enzymes of Clostridium acet-~~
~~butylicum in the fermentation of soybeans.~~
~~Trav. Inst. Chim. Res. Acad. Sci. USSR, Moscow.~~
~~1963, No. 1, p. 103.~~ -- Intensified proteolysis
in mashers with relatively low starch:protein ratio (maize,
wheat, peas) hastens accumulation of vol. N in the mash
and transition to the 2nd fermentation stage, so that yield
of solvents is improved. Amylase activity is intensified by
higher starch:protein ratios (rye, potatoes), causing an
earlier drop in acidity and higher total yield of solvents,
while also favoring alcohol production. *Irina E. Smith*

PM MT

ZALESSKAYA, M.I.

Methods for getting higher butanol output in the acetone-butyl fermentation; theoretical data and experimental results. Trudy TSNIISF no.6:73-81 '58. (MIRA 14:12)

(Butyl alcohol) (Butanol)

ZALESSKAYA, M.I.; LOGOTKIN, I.S.; MARFINA, A.M.; GUS'KOVA, N.P.;
CHEKASINA, Ye.V.

Processing of sugar-beet molasses in the butyl alcohol-acetone
production. Trudy TSNIISP no. 8:52-60 '59. (MIRA 14:1)
(Molasses) (Butyl alcohol) (Acetone)

PEREPIETCHIKOV, Ye.G., dotsent kand.tekhn.nauk; SOLODOVNIKOV, Z.V.;
ZALESSEKAYA, M.P.

Results of the experimental investigation of thermal fields on
surfaces of radiators operating at increased parameters of heat
carriers. Sbor. nauch. trud. Bel. politekh. inst. no.74:10-18
'59. (MIRA 13:8)

(Radiators)

ZALESSKAYA, N.T.

Cavernicolous Chilopoda of the Crimea and the Caucasus. Zool.
zhur. 42 no.7:1022-1030 '63. (MIRA 17:2)

1. Department of Invertebrate Zoology, State University of
Moscow.

KUIMOV, D.T.; ZALESSKAYA, O.M.

Rheumatic fever and multiple sclerosis. Zhur. nevr. i psikh.
64 no.3:370-375 '64. (MIRA 17:5)

1. Klinika nervnykh bolezney (zaveduyushchiy kafedroy -
prof. D.T. Kuimov) Novosibirskogo meditsinskogo instituta.

ZALESSKAYA, S.V.

Viscosity of slags obtained from the reduction smelting of
red mud sinters. Izv. vys. ucheb. zav.; chern. met. 7 no.1:
38-40 '64. (MIRA 17:2)

1. Gor'kovskiy politekhnicheskoy institut.

ZALESSKAYA, S.V.

SOV/137-58-8-18089

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 8, p 268 (USSR)

AUTHOR: Zalesskaya, S. V.

TITLE: Colorimetric Determination of Vanadium in Bauxites and Red Mud (Kolorimetricheskoye opredeleniye vanadiya v boksitakh i krasnom shlamе)

PERIODICAL: Tr. Gor'kovsk. politekhn. in-ta, 1957, Vol 13, Nr 5, pp 90-93

ABSTRACT: In bauxites and red mud V is determined by the colorimetric method based on the production of a P - W - V compound. 1 g of material is fused with 5 g NaOH, the melt is leached out with 50 cc of water and filtered into a 100-cc flask. A 25-cc aliquot is neutralized with 6-N H_2SO_4 to methyl orange with an addition of 3 cc of acid in excess. 0.5 cc of 1% solution of H_2O_2 is added and 4% solution of $KMnO_4$ drop by drop to the appearance of a raspberry-red coloration, which is destroyed 2.5 min later by the addition of a few drops of 1% $NaNO_2$ solution. To the discolored solution are added 1 cc of 15% solution of N_2WO_4 and water to make a total of 50 cc, whereupon the V is determined colorimetrically.

Card 1/1

1. Bauxite—Colorimetric analysis 2. Vanadium— V. N.
Determination

SAMARIN, A.M.; RUDNEVA, A.V.; ZALESSKAYA, S.V.

Effect of the phase composition of slags on the process of cast iron
gravitation in the reduction smelting of red pulp sinters. Izv.vys.
ucheb.zav.; chern.met. 4 no.6:20-26 '61. (MIRA 14:6)

1. Institut metallurgii im. A.A.Baykova.
(Cast iron—Metallurgy) (Slag)

ZALESSKAYA, Yu.M.

Effect of experimental hypo- and hyperthyroidism on the process of the conversion of β -carotene into vitamin A in vitro in guinea pig tissues. Biokhimiia 30 no.6: 1132-1136 N-D '65. (MIRA 19:1)

1. Laboratoriya biokhimii Instituta eksperimental'noy patologii i terapii AMN SSSR, Sukhumi. Submitted November 9, 1964.

PROCESS AND PROPERTIES INDEX																									
1ST AND 2ND DEGREE													3RD AND 4TH DEGREE												
<p>BC</p> <p style="text-align: right;">a-3</p> <p>Hydrogenation of acetylene derivatives. XXII. Dihydrocyclohexylacetylene. J. S. SALIKID (with T. A. KALININA, D. I. ROSANOV, and G. V. TACHENKOV) (J. Gen. Chem. Russ., 1935, 5, 1722-1727).—1:1'-Dihydroxy-1:1'-dicyclohexylacetylene (I) (<i>Ac</i> derivative, b.p. 164-170°, m.p. 45-5-46-5°) and <i>Ar</i> in CHCl_3 afford <i>sp</i>-ditrunc-<i>sp</i>-di-cyclohexylacetylene 1:1'-oxide, m.p. 106-5-107-5°. (I) is readily hydrogenated (Pd) to 1:1'-dihydroxy-1:1'-dicyclohexylacetylene, m.p. 182°, which yields adipic acid when oxidized with KMnO_4. R. T.</p>																									
<p>ASB-1LA METALLURGICAL LITERATURE CLASSIFICATION</p> <p>FROM SYNONYM TO SYNONYM</p> <p>SYNONYM TO SYNONYM</p>																									

1ST AND 2ND CROSS
PROCESSING AND REPRODUCTION UNIT

Molecular transpositions of ketones into ketones under the action of zinc chloride at high temperatures. Al. Favor'skiy, T. P. Zelenyaya, D. I. Rozanov and G. V. Chelmsky. *Russ. J. Chem.* (S. S. R.) 8, No. 12, 1726-1735 (1965). *Chem. Abstr.* 61:14000. When heated at 420° in the presence of ZnCl₂, PhCOCH₂Me, isomerizes to form PhMeCHCOCH₂Me. Similarly, PhCOCHMe, (bp 100°), from PhCH(OH)CHMe, m. 41.5°, by oxidation; semicarbazone, m. 150°) forms PhMeCHCOCHMe, bp 97.8°, which yields a semicarbazone, m. 190.1°, and MePhCHCOCH₂Me, m. 78.9° (oxidation with alkali and H₂). Also, PhCOCHMeEt (bp 112°, d₄ 0.869, d₂₀ 0.873, from PhCH(OH)CHMeEt, m. 22°, by oxidation with Na₂C₂O₄ and H₂SO₄; semicarbazone, m. 127.8°), forms PhMeEtCHCOCHMe, bp 114.16°, d₄ 0.8739, d₂₀ 0.8720, (semicarbazone, m. 177.8°), oxidized to MeEtPhCHCOCH₂Me, m. 84°. Under the same conditions PhCOEt does not isomerize, but undergoes pyrolytic decarboxylation, forming C₆H₆, CO, CO₂, C₂H₄, PhPh, MeCH, CHPh, H₂O, EtCOH, H₂O and resin. Ph cyclohexyl ketone also does not isomerize but forms phenylcyclohexane by elimination of CO. The ability of the ketones to isomerize is explained by the nonplanar and mobility of the O atom. J. F. A.

ASA-51A METALLURGICAL LITERATURE CLASSIFICATION

PROCESSING AND PROPERTIES INDEX	
10	<p>Catalytic rearrangement of ketones. Isomerization of <i>sec</i>-butyl phenyl ketone. T. B. Zalcaskaya. <i>J. Gen. Chem. (U. S. S. R.)</i> 8, 1540-43 (1938).—Previously it was shown that on heating at 320° in the presence of ZnCl₂, <i>o</i>-substituted MeCOPh with 2 and 3 Me groups isomerized to give ketones with an Ac group, while primary ketones (PhCORt) do not isomerize, but undergo pyrolytic decomposition. (cf. Favorskii, <i>et al.</i>, <i>C. A.</i> 30, 2941¹). It was of interest to investigate the behavior of similar ketones with unlike substituents. <i>PhCOCHMeEt</i>, b_p 113.3-14.4°, d₄²⁰ 0.974, n_D²⁰ 1.5087, M. R. 49.68, prepd. from PhCH(OH)CHMeEt, b_p 120-1°, by oxidation with CrO₃, under the same conditions isomerized to PhEtCHAc (I) and not to PhMe-CHCOEt, which theoretically is also possible. I, b_p 105.5-6.5° (semicarbazone m. 191°), oxidized with Br in NaOH soln., gave PhEtCHCO₂H; its amide m. 80°. Thus, the character of the mol. transposition is not influenced by the presence of like and unlike substituents in the ketones. Chas. Blanc</p>
ASAC-3.1A METALLURGICAL LITERATURE CLASSIFICATION	
SEARCH SYMBOL	SEARCH SYMBOL
SEARCH SYMBOL	SEARCH SYMBOL

CHIRIKOVA, L. V.

Mr., Leningrad State Univ. Inst. B. A. I. Gorkon, -1945-.

Mr., Leningrad State Pedagogical Inst. Dr. A. I. Gorkon, -cl943-.

"On the Problem of Catalytic Conversions of Ketones Isomerization of M. A.

-Diethyl-Acetophenone," Zhur. Obshch. Khim., 16, No. 11, 1946;

"The Identification of Alkyl Benzenes," ibid., 17, No. 3, 1947;

"The Problem of Catalytic Conversion of Ketones by Isomerization of Me-

Methyldeoxybenzoin," ibid., 18, No. 6, 1948;

"Problem of the Isomer Conversions of Carbonyl Compounds. The Relation of

Diphenylacetic Aldehyde to Heating with Zinc Chloride," ibid.

Identification of alkylbenzenes. I. Analysis of binary mixtures of acylamine derivatives of *sec*-butylbenzene and isobutylbenzene. I. K. Zaitseva (Leningrad Pedagog. Inst.). *J. Gen. Chem. (U.S.S.R.)* 17, 480 (1947) (in Russian). - Melting ranges were detd. for mixts. of Ac and Bu derivs. of *p*-NH₂ and 2,4-diamino derivs. of *sec*-BuPh and iso-BuPh, as a means of analysis of these binary systems. EtMgBr (from 12 g. Mg and 54 g. EtBr) and 40 g. AcPh in Et₂O gave 40 g. 2-phenyl-2-butanol, b_p 107-8°; this + 10 g. red P and 36 g. III (b. 127°) were heated and slowly treated with 50 g. iodine and, after 9-hr. heating, steam distd. to give 12.3 g. *sec*-BuPh, b_p 70-80°, b₁₀₀ 173-4°. The hydrocarbon (12 g.) was added in 2 min. to 21 cc. concd. H₂SO₄ at -10° with vigorous stirring, and the mixt. was treated slowly with 8 cc. HNO₃ (d. 1.418) and 12 cc. concd. H₂SO₄ at 0-3° over 1 hr.; after stirring 10 min. at 8-10° the mixt. was poured on ice to yield 4 g. *p*-nitro-*sec*-butylbenzene, b_p 104-5°, b_p 120-7°, d₄ 1.0875, d₂₀ 1.0770, n_D 1.53368, and 8 g. 2,4-dinitro-*sec*-butylbenzene, b_p 143-4°, b_p 161-2°, d₄ 1.2411, d₂₀ 1.2195, d₂₀ 1.2294, n_D 1.53338. Reduction of the former with 9 g. Sn and 24 cc. concd. HCl gave 3.1 g. crude amine, which was acylated directly: Ac deriv. (from Ac₂O), m. 125.8-6° (from C₆H₅Cl), m. 126.1° (from 50% EtOH) (I); Bu deriv. (from BuCl), m. 131.0-31.8° (from 80% EtOH) (II). The di-NO₂ deriv. similarly gave 5.9 g. crude diamine after reduction with Sn-HCl, which was directly acylated: di-Ac deriv. (from Ac₂O), m. 191.5° (from 80% EtOH) (III); di-Bu deriv. (from Bu₂Cl), m. 225.6° (from BuOH) (IV). Me₂CO (30 g.) and PhCl₂MgCl (from 52 g. PhCl₂Cl) gave 40

g. dimethylbenzylcarbamid, b_p 199-11°, b_p 21-6°, this was reduced with 10 g. red P and 35 g. III and 55 g. iodine to give 20 g. iso-BuPh, b_p 100-70°, b_p 60-1°; this was nitrated as described above to give 4.5 g. *p*-nitroisobutylbenzene, b_p 125-6°, d₄ 1.0892, d₂₀ 1.0774, n_D 1.5300 and 13.5 g. 2,4-dinitroisobutylbenzene, b_p 154-0°, d₄ 1.2513, d₂₀ 1.2131, n_D 1.54791. Reduction with Sn-HCl gave 2.9 g. crude amine and 5.0 g. (from 8 g. di-NO₂ compd.) crude diamine, resp. Acylation gave *p*-acetylaminobutylbenzene (from Ac₂O), m. 131.5-131.8° (from 30% EtOH) (V); *p*-benzamidobutylbenzene (from BuCl), m. 129.4° (from 80% EtOH) (VI); 2,4-diacetamidobutylbenzene, m. 218° (from 50% EtOH) (VII); and 2,4-dibenzamidobutylbenzene, m. 230.2° (from BuOH) (VIII). The 8 compds. in the powder state were used to make up 4 series of binary mixts.: mono-Ac derivs., di-Ac derivs., mono-Bu derivs., and di-Bu derivs. the m.p.s. of which were detd. by the capillary method. The m.p. curves vs. compn. (iso-Bu vs. *sec*-Bu) are given; this shows the points of beginning and the end of melting of each component pair. These are characterized as follows: I-V shows a narrow melting range at all compns. and has a well-defined min. m.p. of 108-9° at approx. 65% *sec*-BuPh deriv.; II-VI shows a wide melting range at all compns. with a sharp min. at m.p. 107-10° at approx. 40% *sec*-BuPh deriv.; III-VII has a quite narrow melting range at all compns. with a sharp min. of m.p. 182-3.5° at approx. 80% *sec*-BuPh deriv.; IV-VIII has a moderate melting range at all compns. with a very shallow melting diagram with a poorly defined min. at m.p. 210-21° at approx. 35% *sec*-BuPh deriv. The

best derivs. for analysis of mixts. of the 2 butybenzenes
are the mononitrobenzene derivs. G. M. Kosolapoff

COMMON ELEMENTS		PROCESSES AND PROPERTIES INDEX																																																																																																					
Catalytic conversions of ketones. Isomerization of α -methylbenzoylbenzoin. T. E. Zaleskaya. <i>Zhur. Obshch. Khim.</i> (J. Gen. Chem.) 18, 1168-71 (1948). PhMeCHCOPh (II) (21 g.), heated 5 hrs. at 330° with 0.4 mole ZnCl ₂ , underwent isomerization to PhCHCOMe (12 g.). The reaction is conceived as proceeding over a 1st-stage formation, between I and the strong-acid catalyst HA, of an intermediate complex of the form PhMeCHCOPh:OH . . . A. In this complex, the Me group, repelled by Ph, migrates as a result of an electron shift from Me to OH, leading to PhCHCOPhMeOH . . . A, in which the electron shift now occurs from the O atom to the C chain, as a result of which, and owing to the negative electronic effects proper to the Ph group, the latter, rather than the Me group, is repelled from the C=O group, giving PhCHCOMe. N. Tlum		10																																																																																																					
ASAC-55.4 DETAILING LITERATURE CLASSIFICATION																																																																																																							
<table border="1"> <tr> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td><td>9</td><td>10</td><td>11</td><td>12</td><td>13</td><td>14</td><td>15</td><td>16</td><td>17</td><td>18</td><td>19</td><td>20</td><td>21</td><td>22</td><td>23</td><td>24</td><td>25</td><td>26</td><td>27</td><td>28</td><td>29</td><td>30</td><td>31</td><td>32</td><td>33</td><td>34</td><td>35</td><td>36</td><td>37</td><td>38</td><td>39</td><td>40</td><td>41</td><td>42</td><td>43</td><td>44</td><td>45</td><td>46</td><td>47</td><td>48</td><td>49</td><td>50</td><td>51</td><td>52</td><td>53</td><td>54</td><td>55</td><td>56</td><td>57</td><td>58</td><td>59</td><td>60</td><td>61</td><td>62</td><td>63</td><td>64</td><td>65</td><td>66</td><td>67</td><td>68</td><td>69</td><td>70</td><td>71</td><td>72</td><td>73</td><td>74</td><td>75</td><td>76</td><td>77</td><td>78</td><td>79</td><td>80</td><td>81</td><td>82</td><td>83</td><td>84</td><td>85</td><td>86</td><td>87</td><td>88</td><td>89</td><td>90</td><td>91</td><td>92</td><td>93</td><td>94</td><td>95</td><td>96</td><td>97</td><td>98</td><td>99</td><td>100</td> </tr> </table>				1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100				

ZALESSKAYA, T. E.

Concerning isomeric transformations of carbonyl compounds. The relation of diphenylacetic aldehyde on heating with zinc chloride. p. 1172

The isomeric transformation is observed of diphenyl acetic aldehyde into desoxybenzoin after heating with zinc chloride. Hereby precipitates diphenylmethane, which is the product of decomposition of diphenylacetic aldehyde.

The Leningrad Herzen State Pedagogical Institute
June 26, 1947

30: Journal of General Chemistry (USSR) 18 (80) No. 6 (1948)

REMIZOVA, T.B.; ZALESKAYA, T.Ye.

Mechanism of the isomeric transformation of ketones. Par 12:
Action of 72% chloric acid on pivalophenone containing C¹⁴
in carbonyl. Zhur. ob. khim. 34 no 5:1395-1399 My '64.
(MIRA 17:7)

1. Leningradskiy tekhnologicheskii institut tsellyulozno-
bumazhnoy promyshlennosti.

ZALESSKAYA, T.Ye.; REMIZOVA, T.B.

Mechanism of isomeric transformations of ketones. Part 3: Action of perchloric acid and zinc chloride on tert-amyl phenyl ketone. Zhur. ob. khim. 34 no.10:3168-3173 0 '64.

(MIRA 17:11)

1. Leningradskiy tekhnologicheskii institut tsellyulozno-bumazhnoy promyshlennosti.

ZALISSKAYA, T.Ye.; REMIZOVA, T.B.

Mechanism of isomeric conversions of ketones. Part 4: Action of perchloric acid on tert-amyl phenyl ketone containing C^{14} in carbonyl. Zhur. ob. khim. 35 no.1:31-34 Ja '65. (MIRA 12:2)

1. Leningradskiy tekhnologicheskii institut tsellyulozno-bumazhnoy promyshlennosti.

ZALESSKAYA, T.Ye.; REMIZOVA, T.B.

Isomeric conversions of ketones. Part 1: Conversion of pivalophenone
in 67% perchloric acid. Zhur.ob.khim. 33 no.12:3802-3804 D '63.
(MIRA 17:3)

1. Leningradskiy tekhnologicheskii institut tsellyulozno-bumazhnoy
promyshlennosti.

S/079/60/030/007/024/039/XX
B001/B066

AUTHORS: Zalesskaya, T. Ye. and Zhuravl'eva, L. Ye.

TITLE: Identification of Alkyl Benzenes. ¹II. Analysis of Binary Mixtures of Acylamino Derivatives of 2-Phenyl Butane and 3-Phenyl Pentane

PERIODICAL: Zhurnal obshchey khimii, 1960, Vol. 30, No. 7, pp. 2164-2166

TEXT: The purpose of the present work was to identify closely related hydrocarbons that are constituents of a mixture. In order to identify 2-phenyl butane mixed with 3-phenyl pentane, the two hydrocarbons were synthesized and converted to their acylamino derivatives. The thermal analysis of the binary mixtures of the corresponding derivatives was made under conditions applicable with small quantities of the products to be studied. The melting points of mixtures of the benzene derivatives of 2-(p-aminophenyl) butane and 3-(p-aminophenyl) pentane, and of mixtures of the acetyl and benzoyl derivatives of 2-(2,4-diaminophenyl) butane and 3-(2,4-diaminophenyl) pentane were determined. Experimental data (melting points) are given in the following diagrams: Diagram 1: melting Card 1/2 ✓

Identification of Alkyl Benzenes. II. Analysis of Binary Mixtures of Acylamino Derivatives of 2-Phenyl Butane and 3-Phenyl Pentane S/079/60/030/007/024/039/XX B001/B066

points of the binary mixture of the p-benzoyl-amino derivatives of 2-phenyl butane and 3-phenyl pentane; diagram 2: melting points of the binary mixture of the 2,4-diacetyl-amino derivatives of 2-phenyl butane and 3-phenyl pentane; diagram 3: melting points of the binary mixture of the 2,4-dibenzoyl-amino derivatives of 2-phenyl butane and 3-phenyl pentane. Under the experimental conditions, not only the temperature measured when the solid phase disappears, but also the temperature during the appearance of the liquid state is important for characterizing each of the mixtures, as is clearly seen from the melting-point diagrams. There are 3 figures and 4 references: 1 Soviet, 2 US, and 1 German.

ASSOCIATION: Leningradskiy tekhnologicheskii institut tsellyulozno-bumazhnoy promyshlennosti (Leningrad Institute of Technology for the Cellulose and Cotton Industry)

SUBMITTED: June 23, 1959

Card 2/2

ZALESSKAYA, T.Ye.; ZHURAVLJEVA, L.Ye.

Identification of alkyl benzenes. Part 2: Analysis of
binary mixtures of acylamino derivatives of 2-phenylbutane
and 3-phenylpentane. Zhur.ob.khim. 30 no.7:2164-2166
Jl '60. (MIRA 13:7)

1. Leningradskiy tekhnologicheskij institut tsellyulozno-bumazhnoy promyshlennosti.
(Butane) (Pentane)

ZALESSKAYA, T.Ye.

Catalytic conversions of ketones. Zhur.ob.khim. 30 no.7:
2166-2170 J1 '60. (MIRA 13:7)

1. Leningradskiy tekhnologicheskij institut tsellyulozno-
bumazhnoy promyshlennosti.
(Butyrophenone)

ZALESSKAYA, T.Ye.; LAVKOVA, I.K.

Dehydration of α -glycols. Part 1: Effect of the nature of reagents on the course of dehydration of asymmetric methyl p-tolylpinacol. Zhur. org. khim. 1 no.7:1215-1218 J1 '65.

(MIRA 18:11)

1. Leningradskiy tekhnologicheskij Institut tsellyulozno-bumazhnoy promyshlennosti.

ZALESSKAYA, Ye.B.; VOL'FSON, S.I.

Heat hardened Kh5M-U pipes. Mash. i neft. obor. no.4:
26-28 '64. (MIRA 17:6)

1. Gosudarstvennyy nauchno-issledovatel'skiy i proyektnyy
institut neftyanogo mashinostroyeniya.

L 32268-65 EWT(m)/EWP(b)/T/EFA(d)/EWP(w)/EWP(t) MJN/JD/WB

ACCESSION NR: AR4048244

S/0137/64/000/009/I055/I055

SOURCE: Ref. zh. Metallurgiya, Abs. 91345

AUTHOR: Zalesskaya, Ye. B.

TITLE: Properties of 11-13% chromium steels at raised temperatures

CITED SOURCE: Tr. Gos. n.-i. i proyekt. in-t neft. mashinostr.,
vy*p. 2, 1964, 132-137

TOPIC TAGS: chromium steel, temperature effect, heat treatment,
metal mechanical property steel OKh13, steel EI496, steel 1Kh13,
steel 2Kh13

TRANSLATION: For work in media which do not cause metal crystalline
corrosion, steels of brand OKh13 are proposed or steels EI496 (I),
1Kh13 (II), 2Kh13 (III) whose compositions (in %) are: carbon 0.07
(I), 0.12 (II), 0.18 (III); silicon 0.14 (I), 0.33 (II), 0.35 (III);
manganese 0.40 (I), 0.44 (II), 0.44 (III); sulfur 0.016 (I), 0.014
(II), 0.014 (III); phosphorus 0.020 (I), 0.020 (II), 0.015 (III);
chromium 12.83 (I), 13.90 (II), 13.50 (III); nickel 0.56 (I), 0.36

Card 1/2

L 32268-65

ACCESSION NR: AR4048244

(II), 0.54 (III). The σ_{b} , σ_{g} , ψ and a_{k} of the proposed steels were investigated at raised and room temperatures. At 550° σ_{b} (in kg/mm²) is 21.4 (I), 26.1 (II), 31.2 (III); σ_{g} (in kg/mm²) is 13.1 (I), 14.2 (II), 15.6 (III); ψ (in %) is 51.8 (I), 40.0 (II), 32.9 (III); and a_{k} was determined at lowered and at room temperatures after heating for 5000 hrs at 450, 500, 550, and 600°. For I and II, a_{k} is quite suitable. After preheating in the 450-550° range, the a_{k} of III is sharply reduced from 15.2 to 0.3 kg/cm², while at the same time H_{b} is increased from 160 to 230. V. Olenicheva

SUB CODE: MM

ENCL: 00

Card 2/2

ZALESSKAYA, Ye.B.; VOL'FSON, S.I.

Pipes made of 1Kh3VF steel. Stal' 23 no.10:935-936 0 '63.

(MIRA 16:11)

1. Gosudarstvennyy nauchno-issledovatel'skiy proyektnyy institut
neftyanogo mashinostroyeniya.

ZALESSKAYA, Yu.; MARTINSON, E.; TYAKHEPYL'D, L. [Tähepöld, L.]

Glutaminase and asparaginase in the gastric mucosa. Biokhimiia 26
no.2:281-283 Mr-Apr '61. (MIRA14:5)

1. Chair of Biochemistry, State University, Tartu.
(STOMACH) (GLUTAMINASE) (ASPARAGINASE)

ZALESKAYA, Y.M., KHOLLO, V.I., VILLAKO, I. A., (USSR)

"Biosynthesis of Hexosamines in the Gastric Mucosa in Connection
with Ammonia Conversions in it."

Report presented at the 5th Int'l. Biochemistry Congress, Moscow,
10-16 Aug. 1961.

ZALESSKAYA, Yu.M.; MARTINSON, E.E.; TYAKHEPYL'D, L.Ya. [Tahepold, L.]

Effect of vitamin C on the synthesis of glutamine and amidiza-
tion of proteins in the brain. Vop. pit. 22 no.3:60-64
My-Je '63. (MIRA 17:8)

1. Iz kafedry biokhimii (zav. - prof. E.E. Martinson) Tartuskogo
universiteta.

ZALESKAYA, Yu.M.; MARTINSON, E.E.; TYAKHEFYL'D, L.ia. [Tah; old, L.]

Disorders of glutamine synthesis and amidation of brain proteins
in vitamin C deficiency. Vop. pit. 23 no.1:17-21 Ja-F '64.
(MIP4 17:8)

1. Iz kafedry biokhimii (zav. - prof. E.E. Martinson)
Tartuskogo universiteta.

ZALESSKAYA, V. V.; FILATOVA, Z. V.; RUBEL, N. N.; TIKHONOVA, V. I.;
SOFRONOV, B. N.; PETROPAVLOVSKAYA, N. A.; SMIRNOVA, A. M.

"Special features of the microbiological immuno-epidemiological
characteristics of scarlet fever treated with penicillin."

Report submitted at the 13th All-Union Congress of Hygienists,
Epidemiologists and Infectionists. 1959

SALETSKAYA, Y. V.

"Fluorographic Examinations of School Children of the Dzerzhinsk Region of
Moscow Oblast," Prob. Tuber., No. 3, 1949.

Mbr., Roentgen & Dispensary Dept., Moscow Oblast Sci. Res. Tuberculosis Inst.,

-c1949-.

L 25807-65 EWT(a)/EWP(b)/T/EWI(d)/EWP(w)/EWP(t) MJW/AD
 ACCESSION NR: AR5605028 S/C277/54/000/012/0013/0013 22
 17

SOURCE: Ref zh. Mashinostroitel'nyye materialy, konstruktai i raschet dotaloy mashin. Otd. vyp., Abs. 12.48.84 B

AUTHOR: Zalesskaya, Ye. B.

TITLE: The high temperature properties of 11-13% chrome steels 18

CITED SOURCE: Tr. Gos. n.-i. i proyekt. in-t neft. mashinostr., 1964, vyp. 2, 132-137

TOPIC TAGS: metal mechanical property, chromium steel/ EI496 steel, Okh13 steel, 1Kh13 steel, 2Kh13 steel 16 18

TRANSLATION: The σ_b , σ_g , ψ , a_k , and σ_{dp} of OKh13 (EI496), 1Kh13 and 2Kh13 steels are studied at high (100-650°) and room temperatures on the basis of 1000, 10,000 and 100,000 hours. At 550°, the steels have respectively: σ_b =24.1, 26.1, 31.2; σ_g =13, 14.2, 15.6 kg/mm² and ψ =51.8, 40, 32.9%. The a_k was determined at low temperatures (down to -40°) and when heated for 5000 hours at 450, 500, 550, 600 and 650°. For the first two steels, the a_k was fully acceptable. After heating in the 450-500° temperature range, the a_k of 2Kh13 steel is reduced from 16.9 to 0.6 kg/cm² at 20° while the hardness is simultaneously increased (from HB168 to Card 1/2

L 25807-65

ACCESSION NR: AR5005028

HB230). It is recommended that the temperature at which 0Kh13 and 1Kh13 steels are used should be increased from 540 to 550° in media which do not cause inter-crystalline corrosion.

ENCL: 00

SUB CODE: MM

Card 2/2

L 23365-65 EWT(m)/EWA(d)/EWP(t)/EWP(b) MJW/JD/WE

ACCESSION NR: AR5000736

S/0277/64/000/009/0009/0009

SOURCE: Ref. zh. Mashinostroitel'nyye materialy*, konstruktii i raschet detaley mashin. Gidroprivod. Otd. vyzp., Abs. 9.48.58

AUTHOR: Zalesskaya, Ye. B.; Vol'fson, S. I.

TITLE: Corrosion resistant pipes of Kh8VF for furnaces and connecting piping in oil refineries

CITED SOURCE: Tr. Gos. n.-i. i proyekt. in-t neft. mashinostr., vyzp., 2, 1964, 126-131

TOPIC TAGS: pipe, metal corrosion, corrosion resistance, sulfur, oil refining/ steel Kh8VF, steel Kh5M, steel Kh5VF

TRANSLATION: The results of an investigation of mechanical properties during short term elongation at temperatures of 20-600°, impact strength at temperatures from 20 to -40°, and long term strength and creep at temperatures of 500-650° are presented for steels Kh8VF, Kh5M, and Kh5VF. All the steels have almost identical properties. Actual use in furnaces of oil refineries showed that the resistance of

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ACCESSION NR: AR5000736

pipes made of steel Kh8VF in oil media containing sulfur was not less than two times greater than the resistance of pipes made of steels Kh5M and Kh5VF, whose resistance was about the same. There is a marked economic advantage in using pipes made of steel Kh8VF. 2 figures. 6 tables.

SUB CODE: MM

ENCL: 00

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PA 15/49T27

USSR/Electricity
Terminology

Jul 48

"For the Accuracy of Electrical Engineering Terms,"
Prof A. M. Zalesskiy, Dr Tech Sci, Leningrad Polytech
Inst imeni Kalinin, 1 p

"Elektrichestvo" No 7

Recommends setting up commission on electrical termin-
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Technical periodicals must also help.

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SA 3224. Investigation of the impulse strength of insulation in electrical machinery. ZALESSKI, A. M. AND ZINSKIRMAN, A. S. *Elektrichestvo* (No. 2) 9-13 (Feb., 1949) In Russian. Tests were made on a number of representative samples and the results subjected to statistical analysis. It was found that the impulse breakdown voltage displayed a wide dispersion, indicating that the average standard of impulse strength could be considerably raised by greater care in manufacturing processes. At room temperatures the impulse coefficient (ratio of impulse breakdown voltage to that at 50 c/s) was a little > 1 for sloping waves (17 kV/ μ s) but was about 15% higher for steep waves (107 kV/ μ s.) The coefficient was considerably greater at the working temperature of 105°C. A continuation of the tests is considered desirable, using a large number of specimens to study the relationship of breakdown voltage to the number of repetitions of the impulse and also to investigate the effect of ageing of the insulation on its breakdown strength. B. H.

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USSR/Electricity - Circuit Breakers Sep 51

"Testing of the 110-KV, 2,500-MVA Air Circuit Breaker VV-110 for Interrupting Capacities up to 3,000 Megavolt-Amperes," A. M. Zaleskiy, LPI (Leningrad Polytech Inst Imeni Kalinin)

"Elektrichestvo" No 9, pp 10,11

Cites results of tests of the interrupting capacity of the type VV-110 air breaker, now being produced by the "Elektroapparat" plant. The tests were made in the Dept of Elec Equipments, Lab of High-Voltage Techniques, pressures of 15-22 atm

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USSR/Electricity - Circuit Breakers Sep 51
(Contd)

and powers up to 3,100 megavolt-amp were used in the tests. Submitted 9 Mar 51.

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